Catalytic Oxidation of CW Agents Using H_2O_2 in Ionic Liquids

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Report Documentation Page

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Structures of ionic liquids

cations:

anions: BF₄-, PF₆-, SbF₆-, NO₃-, CF₃SO₃-, (CF₃SO₃)₂N-, ArSO₃-, CF₃CO₂-, CH₃CO₂-, Al₂Cl₇-



Common Cationic Surfactants

$$H_3C(H_2C)_{14}H_2C$$
 CH_3
 CH_3
 CH_3

R R Br

CTAB

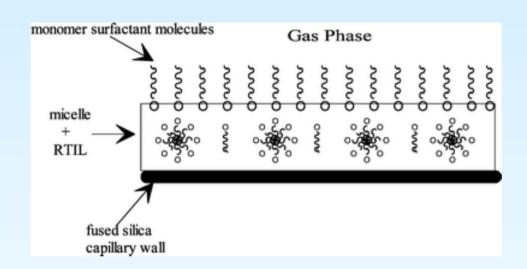
morpholinium salts

N-alkylpyridinium halides

salts of alkyl-substituted pyridine

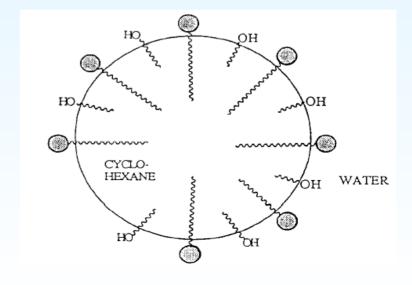


Micelles to Microemulsions





Rapid, Cheap Selective





Microemulsion Formulations

- ◆ 69% H₂O
- ◆ 9% EtOH
- ◆ 11% [BMIM] BF₄
- * 11% Cyclohexane
 o/w

- ◆ 5% H₂O
- ◆ 10% EtOH
- **◆ 5% [BMIM] PF**₆
- * 80% Cyclohexane w/o

Structures of HD, VX, GB, and GD

(1) HD, mustard; (2) VX; (3) GB or Sarin; (4) GD or Soman

Decontamination of Mustard and Phosphorous (V) Esters

PhSCH₂CH₂CI
$$\frac{H^+, H_2O_2}{H_2O}$$
 PhSCH₂CH₂CI $\frac{Ph}{O}$ + EtSO₃⁻ R SEt

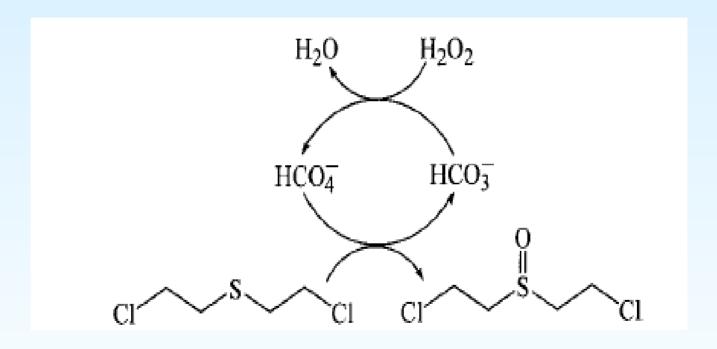
Hydrogen Peroxide with added carbonate

CICH₂CH₂SEt
$$\frac{H_2O_2 / Me_4NHCO_3}{[BMim] BF_4} \sim CICH_2CH_2SEt$$

- Less than 2 minutes
- 2 mole equiv. Of H_2O_2

Comp (EtOH): Marques et al, J. Org. Chem. 2001, 66, 7588-7595

Hydrogen peroxide, bicarbonate, and organic cosolvents afford rapid, decontamination of CWA



Wagner, G. W.; Yang, Y.-C. *Ind. Eng. Chem. Res.* 2002, *41*, 1925 -1928

Solubility in a Microemulsion vs glycol/water

SO	luhi	lizate	mio
<u>30.</u>	luvi	IIZatt	11111

croemulsion 3:1 PG/water

paraoxon

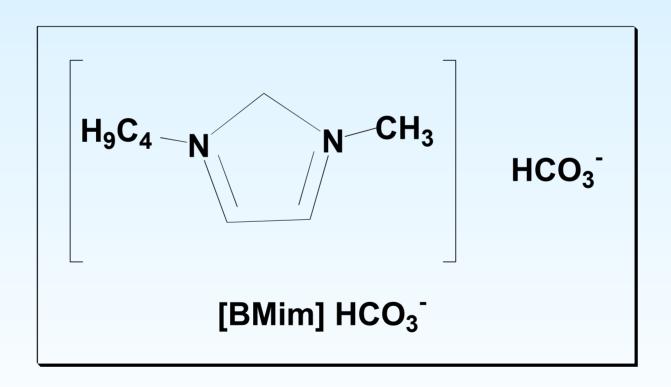
470 mg/0.5 mL 215 mg/0.5 mL

half-mustard

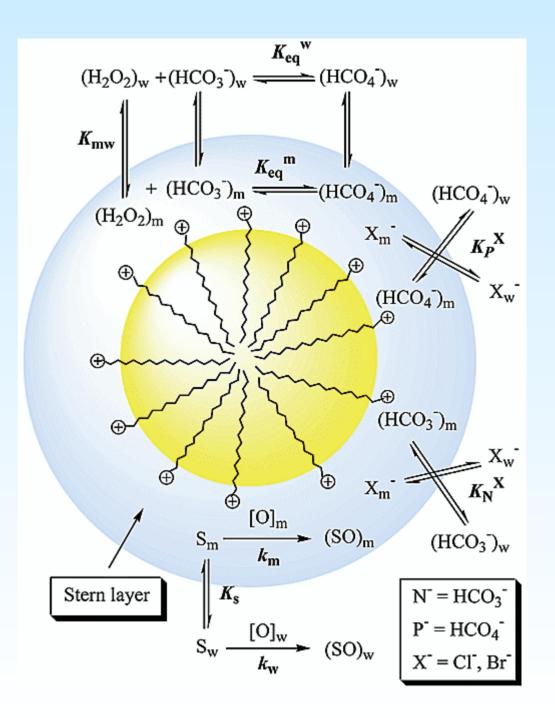
75 mg/0.5 mL

b

Bicarbonate-activated Peroxide (BAP)



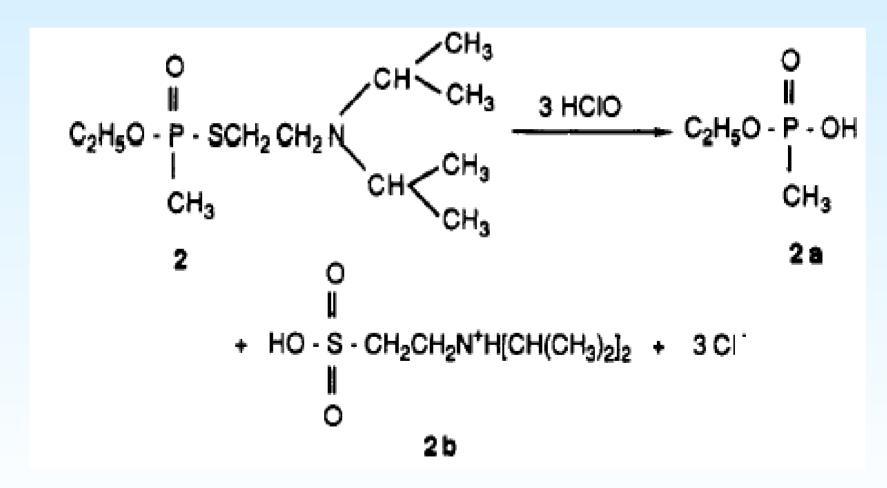
BAP oxidations in aqueous cationic micelles



Oxidation of sulfides to sulfoxides and sulfones with 30% H_2O_2

Sato, K. et al *Tetrahedron* 2001, 57, 2469-2476

VX Decontamination by Bleach



Yang, Y.-C. et al. *Chem. Rev.* **1992**, *92*, 1729-1743

Rapid detoxification of HD using magnesium monoperoxyphthalate (MMPP)

$$CO_{3}H_{i_{M_{1}M_{1}}} \underbrace{Mg^{2+}}_{i_{M_{1}M_{1}}} \underbrace{HO_{3}C}_{i_{M_{1}M_{1}}} \underbrace{Mg^{2+}}_{i_{M_{1}M_{1}}} \underbrace{OOC}_{OOC} \underbrace{OOC}_{OOC$$

Gonzaga, F. et al., New Journal of Chemistry 2001, 25, 151-155

Mustard oxidation in the microemulsion

- Reaction capacity is excellent
- Sulfoxide is formed quantitatively
- Microemulsions prepared from anionic, non-ionic, and cationic surfactants

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Wisdom on the journey...



He who can no longer pause to wonder and stand rapt in awe is as good as dead; his eyes are closed.

Albert Einstein